

# A Wittig Reaction Involving a Novel Rearrangement: Confirmation by X-Ray Crystallography

By E. G. BRAIN, F. CASSIDY, and A. W. LAKE\*

(Beecham Research Laboratories, Medicinal Research Centre, Harlow, Essex)

and P. J. COX and G. A. SIM\*

(Chemistry Department, University of Glasgow, Glasgow G12 8QQ)

**Summary** Reaction of the cyclopentanone (1) with the Wittig reagent ethylenetriphenylphosphorane gives as product not the expected ethylenecyclopentane but the ethylenecyclohexane (3); structures of the alcohol (2), the lactone (4a), and the ether (5a) were established by X-ray diffraction methods.

enylophosphonium bromide. The product is the cyclohexane derivative (3) and not the expected ethylenecyclo-

DURING recent studies in the synthesis of secosteroids we

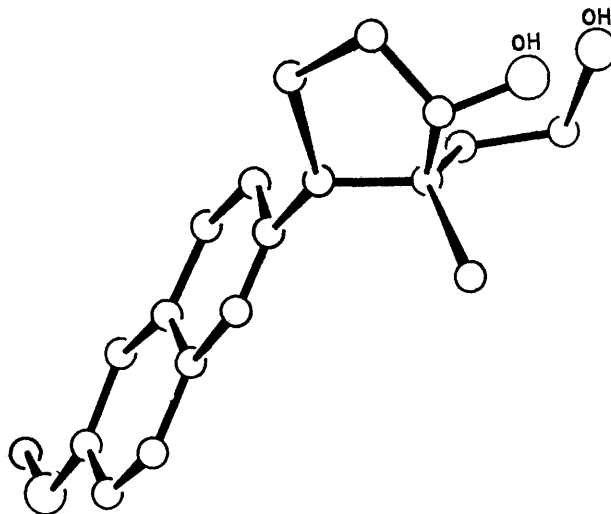
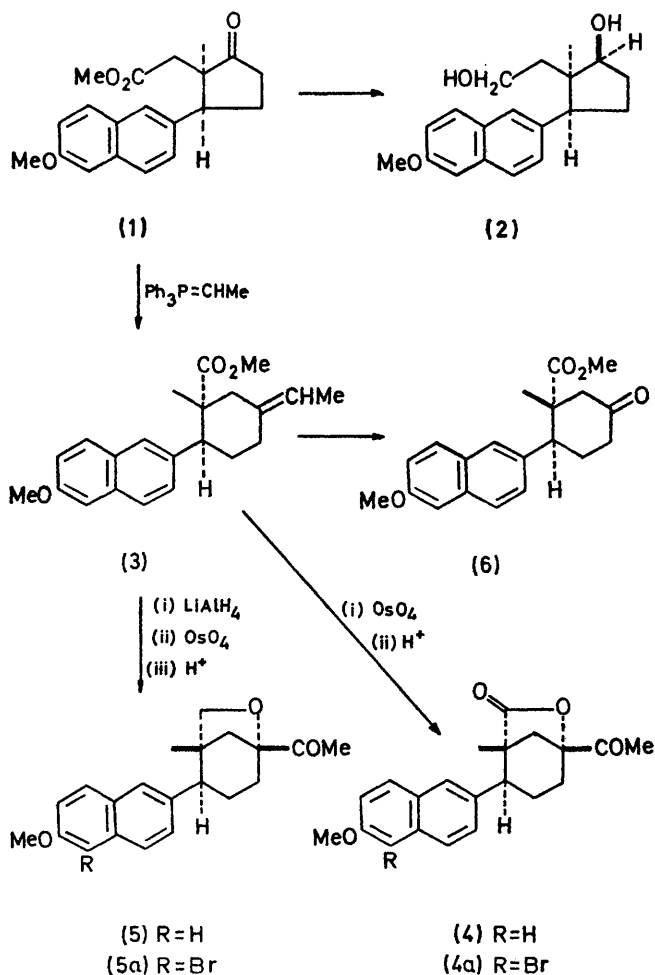


FIGURE 1. Molecular structure of the alcohol (2).

pentane. The analogous reaction between androst-17-ones and the ethyldene Wittig reagent, on the other hand, is known to yield the expected pregn-17(20)-enes.<sup>2</sup> The mechanism of the rearrangement is by no means clear; several possibilities have been considered but none has much in the way of precedent.

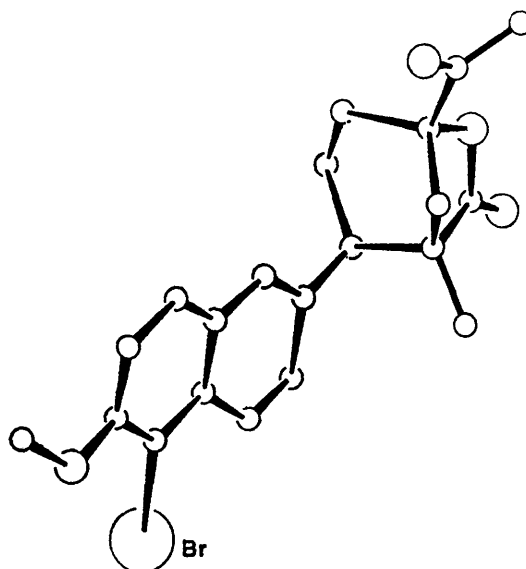


FIGURE 2. Molecular structure of the lactone (4a).

prepared compounds (1) and (2) and some related cyclopentanes, as ( $\pm$ )-mixtures.<sup>1</sup> The relative stereochemistries of these compounds were confirmed by a three-dimensional X-ray analysis of (2) (see Figure 1). Further work in this series has revealed a novel rearrangement during the Wittig reaction of (1) with a five-fold excess of ethylenetriphenylphosphorane generated by the action of sodium methylsulphinylmethide ( $\text{MeSOCH}_2^- \text{Na}^+$ ) on ethyltriph-

Oxidation of (3) with osmium tetroxide-triethylamine oxide peroxide followed by acid work-up gave the lactone (4). Reduction of (3) with lithium aluminium hydride, followed by oxidation with osmium tetroxide-triethylamine oxide peroxide and acid cyclization gave the ether (5). (4) and (5) were treated with bromine at  $-20^\circ$  in the presence of anhydrous sodium carbonate and magnesium sulphate in dry chloroform as solvent to give the mono-bromo-derivatives (4a) and (5a), the structures and relative stereochemistries of which were fully characterized by three-dimensional X-ray analyses (see Figures 2 and 3). Treatment of the Wittig product (3) with osmium tetroxide and sodium chlorate followed by cleavage of the resultant vic-glycol with sodium metaperiodate did not give the cyclopentanone (1) but, instead, a compound having spectroscopic properties consistent with structure (6) (e.g.  $\nu_{\text{CO}}$   $1715\text{ cm}^{-1}$ , absence of  $\text{CH}_2\text{CO}_2\text{Me}$  signal in n.m.r. spectrum).

The alcohol (2) crystallizes in the monoclinic space group  $P2_1/c$  with  $Z = 2$  and cell dimensions  $a = 17.010(10)$ ,  $b = 7.264(5)$ ,  $c = 13.368(7)\text{ \AA}$ ,  $\beta = 93.52(4)^\circ$ . The lactone (4a) crystallizes in the triclinic space group  $P\bar{1}$  with  $Z = 2$  and cell dimensions  $a = 12.056(5)$ ,  $b = 13.026(5)$ ,  $c = 7.593(3)\text{ \AA}$ ,  $\alpha = 90.38(3)$ ,  $\beta = 106.07(3)$ ,  $\gamma = 124.42(3)^\circ$ . The ether (5a) crystallizes in the triclinic space group  $P\bar{1}$  with  $Z = 2$  and cell dimensions  $a = 12.076(6)$ ,  $b = 13.090(5)$ ,  $c = 7.490(3)\text{ \AA}$ ,  $\alpha = 92.65(5)$ ,  $\beta = 104.90(5)$ ,  $\gamma = 124.55(5)^\circ$ . X-Ray intensity data for all three compounds were measured with  $\text{Mo-K}_\alpha$  radiation on a computer-controlled four-circle diffractometer. The crystal structure of (2) was solved by direct methods, and the crystal structures of (4a) and (5a) were solved by the heavy-atom

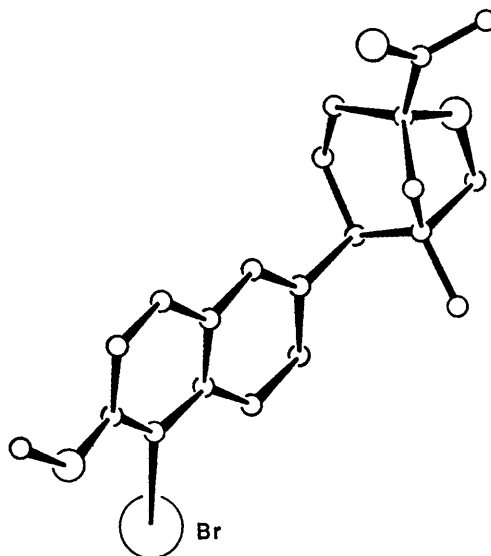


FIGURE 3. Molecular structure of the ether (5a).

procedure. The atomic co-ordinates were refined by least-squares calculations, and current values of  $R$  are 0.077 over 2006 reflections for (2), 0.055 over 2928 reflections for (4a), and 0.091 over 2330 reflections for (5a).

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<sup>1</sup> E. G. Brain, F. Cassidy, M. F. Constantine, J. C. Hanson, and D. J. D. Tidy, *J. Chem. Soc. (C)*, 1971, 3846.

<sup>2</sup> A. M. Krubiner and E. P. Oliveto, *J. Org. Chem.*, 1960, **31**, 24; G. Drefahl, K. Ponsold, and H. Schick, *Chem. Ber.*, 1965, **98**, 604.